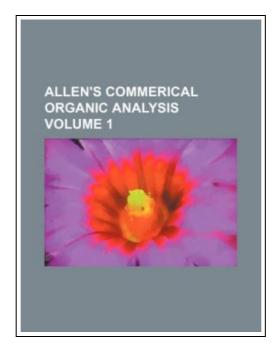
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Rarebooksclub.com, United States, 2012. Paperback. Book Condition: New. 246 x 189 mm. Language: English . Brand New Book \*\*\*\*\* Print on Demand \*\*\*\*\*\*This historic book may have numerous typos and missing text. Purchasers can download a free scanned copy of the original book (without typos) from the publisher. Not indexed. Not illustrated. 1918 Excerpt: .an adulterant of iodoform (Pharm. Jour. 3, 1883, 14, 493). It may be detected by agitating the sample with dilute solution of sodium carbonate, carefully neutralising the nltrate with acetic acid, and adding potassium nitrate, when a yellow low precipitate of the sparingly soluble potassium picrate will be thrown down. The iodoform may also be separated by treating the sample with sodium hydroxide solution and agitating the liquid with chloroform, when only the picric acid will remain in the aqueous liquid. Picric acid may also be detected by the reddish-brown colouration produced on heating the cold aqueous solution of the sample with potassium cyanide. SUGARS. E. FRANKLAND ARMSTRONG, D. Sc., Ph. D.F. C. G. I. Under the generic name of sugars is included a large number of substances occurring naturally in the animal or vegetable kingdoms or produced from the so-called glucosides by the action of ferments or dilute acids. The sugars constitute a group of closely-allied compounds, in many cases distinguishable from each other only with considerable difficulty, while their quantitative separation is frequently impossible in the present condition of chemistry. As a class, the sugars are crystallisable, readily soluble in water, somewhat less soluble or wholly insoluble in alcohol, and insoluble in ether and other solvents immiscible with water. A sweet taste is possessed by nearly all sugars to a greater or less extent. Glycerol and glycol have a sweet taste, and, like the sugars, are polyatomic alcohols. In many cases the sugars exert...



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