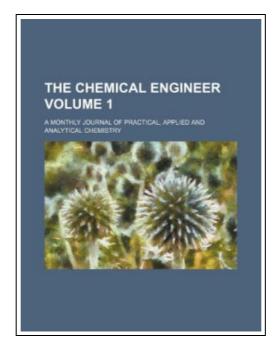
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Rarebooksclub.com, United States, 2012. Paperback. Book Condition: New. 246 x 189 mm. Language: English . Brand New Book ***** Print on Demand ******This historic book may have numerous typos and missing text. Purchasers can download a free scanned copy of the original book (without typos) from the publisher. Not indexed. Not illustrated. 1905 edition. Excerpt: .should have a bulk of about 130 c. c. and be free from ammonium nitrate, chlorides, or free nitric acid (last, except in minute traces). To it are added 1.5 to 2 c. c. of commercial formalin, its temperature raised to 60-65C and the copper deposited by a current of 2.6 to 3.0 amperes at 2 volts. When the deposition is accomplished as ascertained by testing a small portion for copper, the deposit is washed with water without interrupting the current, then with 95 alcohol and finally with ether, and then dried over sulphuric acid in vacuo. The analysis takes from 2 to 2\$ hours. The cathode is a platinum dish and the anode a spiral of the same metal kept as deep as possible in the solution, and passing through an inverted funnel, which just touches the liquid, to avoid loss by spray. THE PRODUCTION AND MODERN USES OF CARBONIC ACID. By JOHN C. MIHOR, Jr., Secretary of the N. Y. Carbonic Add Oai Co. The manufacture of liquefied carbonic acid gas is an industry of comparatively recent origin. First prepared by Faraday under experimental conditions, the liquefied gas remained a chemical curiosity until the late seventies. By that time the problem of manufacturing both the machinery to compress the gas and tubes capable of holding it safely for shipment had been solved, thus insuring the permanence of the industry and its rapid growth. To the pioneers of this industry must be given credit...



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